

An Expedient Synthesis of *N*-Acceptor-Substituted 2,3-Dihydropyrrols from the Corresponding 2-Pyrrolidinones

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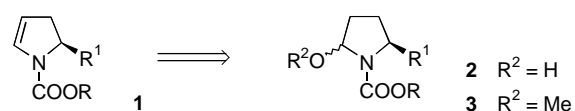
Dedicated to Prof. Dr. F. Vögtle on the Occasion of his 60th Birthday

Keywords: Nitrogen heterocycles, Synthetic methods, Eliminations, Reductions, Dihydropyrrols

Abstract. The title compounds **1** were prepared from the corresponding *N*-acceptor substituted 2-methoxypyrrolidines **3** by elimination with trimethylsilyl trifluoromethanesulfonate (TMSOTf) and *N,N*-di-*iso*-propyl ethyl amine (6 exam-

ples, 57–84% yield). The enantiomerically pure *N*-methoxycarbonyl protected elimination substrates **3a** and **3ab** were synthesized from the L-pyrroglutamic acid derived pyrrolidinones **6a** and **6b** in three steps (80–83% yield overall).

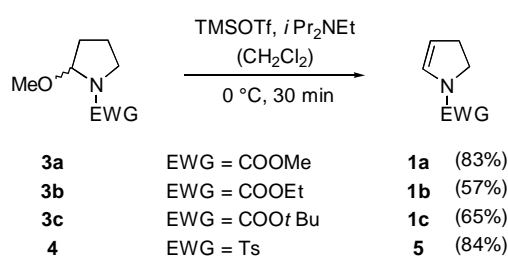
N-Alkoxy carbonyl protected chiral 2-substituted 2,3-dihydropyrrols **1** are heterocyclic olefines the synthetic potential of which has been explored only to a limited extent. There are studies on addition reactions to a proline-derived dihydropyrrol ($R^1 = \text{COOMe}$) and their facial diastereoselectivity [1, 2]. A 2-alkenyl substituted 2,3-dihydropyrrol was used in racemic form as an entry into biologically active *aza*-heterocycles [3]. Recent interest from our group centered on the ability of 2,3-dihydropyrrols to react as alkene components in the Paternò-Büchi reaction [4].



Scheme 1

If R^1 is a vinyl or aryl substituent the enantiomerically pure dihydropyrrols **1** can be prepared from unsubstituted *N*-alkoxycarbonyl-2,3-dihydropyrrols by an enantioselective Heck reaction [5]. If R^1 is an alkyl substituent the target compounds **1** are ideally prepared from the corresponding *N,O*-acetals **3** or from the lactamols **2** which may in turn be obtained either oxidatively from pyrrolidines [1, 6] or reductively from pyrrolidinones [7]. The oxidative approach has been reported to be not regioselective in some instances [8]. The proton-catalyzed [1] and the uncatalyzed [7] thermal elimination procedures which have been used to transform the *N*-alkoxycarbonyl-pyrrolidines **2** or **3** into dihydropyrrols did not prove to be well suited for small scale work in our hands. In this note we report on a new method for the conversion of chiral and achiral pyrrolidines **3** to the title compounds. In addition, we describe a reductive approach to the chiral *N,O*-acetals **3** ($R^1 = \text{alkyl}$) starting from pyrroglutamic acid-derived pyrrolidinones.

It was assumed that the *N*-alkoxycarbonyl iminium ion formation with TMSOTf which has been well documented [9] can serve as an entry into the projected elimination of MeOH from **3** if it was combined with non-nucleophilic base treatment. Indeed, the reagent combination trimethylsilyl trifluoromethanesulfonate (TMSOTf) and *N,N*-di-*iso*-propyl ethyl amine (Hünig base) previously employed for the elimination of *O,O*-acetals to enol ethers [10] proved useful for the preparation of the pyrrolines **1** and **5** from the corresponding 2-methoxy pyrrolidines **3a–c** [11] and **4** [12] as depicted in Scheme 2.

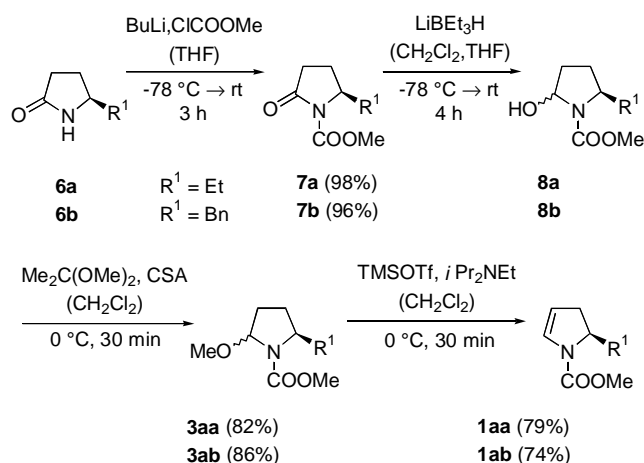


Scheme 2

Gratifyingly, the *N*-methoxycarbonylprotected pyrrolidine **3a** reacted particularly well and yielded the dihydropyrrol **1a** in 83% yield. Previous experiments had revealed that the *N*-methoxycarbonyl protected substrates are superior to other *N*-protected dihydropyrrols in the Paternò-Büchi reaction [13]. Upon photochemical reaction with benzaldehyde **1a** gave the corresponding oxetane in 56% yield whereas the yields with **1b** and **1c** were only 41% and 52%. The *N*-tosyl protected substrate **5** did not react at all.

The extension to chiral dihydropyrrols was consequently carried out with *N*-methoxycarbonyl pyrrolidinones which bear an alkyl group adjacent to the nitrogen atom. Starting

from *L*-pyroglutamic acid the lactams **6** can be readily prepared [14]. After protection of the nitrogen atom the reduction of the pyrrolidinones **7** was conducted with LiEt_3BH following the procedure of Dieter and Sharma [7] (Scheme 3). Subsequent acetalization of the lactams **2aa** and **2ab** to the *N,O*-acetals **3aa** and **3ab** proceeded best with dimethoxypropane and camphor sulfonic acid (CSA) in CH_2Cl_2 as the solvent [15]. The generality of the elimination procedure was finally demonstrated by the smooth conversion of the *N,O*-acetals to the corresponding dihydropyrrols **1aa** and **1ab**.



Scheme 3

In summary, the conversion of the readily available *N*-acceptorsubstituted 2-methoxypyrrolidines **3** to 2,3-dihydropyrrols **1** can be advantageously carried out with TMSOTf and *N,N*-di-*iso*-propyl ethyl amine. Chiral 2-methoxypyrrolidines such as **3aa** and **3ab** derived from *L*-pyroglutamic acid are well suited for the elimination protocol and the method described above gives ready access to the title compounds **1**. The very same procedure has recently been used to prepare a 2-nonyl substituted 2,3-dihydropyrrol which was converted to (+)-preussin [4b]. Further applications of the dihydropyrrols **1** in cycloaddition reactions are currently being investigated in our laboratories and will be reported in due course.

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Experimental

All reactions involving water sensitive chemicals were carried out in flame-dried glassware with magnetic stirring under Ar. Chemicals and solvents for this kind of reactions were distilled from an appropriate drying agent. Common solvents (cyclohexane, ethyl acetate, pentane, *tert*-butyl methyl ether) used for chromatography were distilled prior to use. The 2-methoxypyrrolidines **3a** [1], **3b** [11b], **3c** [11c], and **4** [12] and the pyrrolidinones **6a** [14a] and **6b** [14b] were prepared as described previously. All other reagents and solvents were

used as received. – IR: Nicolet 510M FT-IR. – MS: Varian CH 7. – ^1H and ^{13}C NMR: Bruker AC-300. Chemical shifts are reported relative to tetramethylsilane as an internal reference. CDCl_3 was used as solvent unless noted otherwise. – Elemental Analyses: Varian Elementar vario EL. – TLC: glass-backed plates (Merck 0.25 mm silica gel 60-F); eluent given in brackets, a cyclohexane (CH)/ethyl acetate (EA) or a pentane (PE)/*tert*-butyl methyl ether (TBME) mixture was used unless stated otherwise; detection by UV or by coloration with ceric ammonium molybdate (CAM). – Flash chromatography [16] (FC): Merck silica gel 60 (230-400 mesh) (50 g for 1 g of material to be separated).

N-Methoxycarbonyl-2,3-dihydropyrrol (**1a**)

Procedure A. 9.0 mmol of TMSOTf (1.96 g, 1.70 ml) was added dropwise to an ice-cooled solution of 7.5 mmol of 2-methoxypyrrolidine **3a** [1] (1.20 g) and 9.7 mmol of *N,N*-di-*iso*-propyl ethyl amine (1.28 g, 1.70 ml) in 20 ml of CH_2Cl_2 . After 30 min the reaction was quenched by adding 40 ml of pentane. The reaction mixture was filtered through a Celite pad and concentrated under reduced pressure. The crude product was purified by flash chromatography (PE/TBME = 90/10). Yield 800 mg (83%). The analytical data for this compound were in accord with those reported in the literature [6b].

N-Ethoxycarbonyl-2,3-dihydropyrrol (**1b**)

Following the protocol described in procedure **A** (*vide* above) 0.65 mmol of 2-methoxypyrrolidine **3b** [11b] (112 mg) were transformed into the desired product. Purification by flash chromatography (PE/TBME = 90/10). Yield 52 mg (57%). The analytical data for this compound were in accord with those reported in the literature [6b].

N-*tert*-Butoxycarbonyl-2,3-dihydropyrrol (**1c**)

Following the protocol described in procedure **A** (*vide* above) 6.5 mmol of 2-methoxypyrrolidine **3c** [11c] (1.31 g) were transformed into the desired product. Purification by flash chromatography (PE/TBME = 90/10). Yield 712 mg (65%). The analytical data for this compound were in accord with those reported in the literature [6b].

N-*p*-Toluenesulfonyl-2,3-dihydropyrrol (**5**)

Following the protocol described in procedure **A** (*vide* above) 2.35 mmol of 2-methoxypyrrolidine **4** [12] (600 mg) were transformed into the desired product. Purification by flash chromatography (CH/EA = 95/5 → 90/10). Yield 440 mg (84%). The analytical data for this compound were in accord with those reported in the literature [12].

(*S*)-*N*-Methoxycarbonyl-5-ethyl-2-pyrrolidinone (**7a**)

Procedure B. A 1.47M solution of *n*-butyl lithium in *n*-hexane (7.5 ml, 11 mmol) was added dropwise to a cooled solution of 10 mmol of pyrrolidinone **6a** [14a] (1.13 g) in 20 ml of THF at $-78\text{ }^\circ\text{C}$. The mixture was stirred for 1 h at this temperature. 14.3 mmol of methyl chloroformate (1.35 g, 1.10 ml) were added, and the reaction mixture was allowed to warm slowly to room temperature. After 3 h it was quenched

with 15 ml of a saturated aq. NH_4Cl solution. The aqueous layer was extracted with TBME (3 × 20 ml). The combined organic layers were dried with MgSO_4 , filtered and concentrated *in vacuo*. The crude product was purified by flash chromatography (TBME). Yield 1.68 g (98%). – $R_f = 0.44$ (EA). – $[\alpha]_D^{25} = -104.9$ ($c = 1.1$, CHCl_3). – IR (film): $\nu/\text{cm}^{-1} = 2965$ (m), 1790 (vs, C=O), 1750 (s), 1720 (s), 1305 (m). – ^1H NMR (300 MHz): $\delta/\text{ppm} = 0.95$ (t, $^3J = 7.5$ Hz, 3H, CH_2CH_3), 1.41–2.11 (m, 4H, CH_2CHCH_2), 2.35 (ddd, $^2J = 17.8$ Hz, $^3J = 9.5$ Hz, $^3J = 1.6$ Hz, 1H, COCHH), 2.52 (ddd, $^2J = 17.8$ Hz, $^3J = 9.2$ Hz, $^3J = 1.7$ Hz, 1H, COCHH), 3.76 (s, 3H, OCH_3), 4.04 (tdd, $^3J = 8.8$ Hz, $^3J = 3.3$ Hz, $^3J = 1.9$ Hz, 1H, CHN). – ^{13}C NMR (75.5 MHz): $\delta/\text{ppm} = 9.4$ (CH_2CH_3), 21.8 (COCH_2CH_2), 26.1 (COCH_2), 31.2 (CH_2CH_3), 53.1 (OCH_3), 59.1 (CHN), 152.0 (COO), 173.8 (CON). – MS (EI), m/z (%): 171 (10) [M^+], 142 (100) [$\text{M}^+ - \text{C}_2\text{H}_5$], 98 (74) [$\text{C}_6\text{H}_{12}\text{N}^+$], 41 (87) [$\text{C}_2\text{H}_3\text{N}^+$]. – HRMS Calcd. for $\text{C}_8\text{H}_{13}\text{NO}_3$: 171.0895. Found: 171.0899.

$\text{C}_8\text{H}_{13}\text{NO}_3$ Calcd: C 56.13 H 7.65 N 8.18
(171.09) Found: C 56.38 H 7.79 N 8.15.

(S)-N-Methoxycarbonyl-5-benzyl-2-pyrrolidinone (7b)

Following the protocol described in procedure B (*vide above*) 1.40 mmol of 2-pyrrolidinone **6b** [14b] (245 mg) were transformed into the desired product. Purification by flash chromatography (PE/TBME = 40/60). Yield 313 mg (96%). – $R_f = 0.38$ (TBME). – $[\alpha]_D^{25} = -87.1$ ($c = 7.9$, CHCl_3). – IR (film): $\nu/\text{cm}^{-1} = 2955$ (m), 1715 (vs, C=O), 1750 (s), 1790 (vs), 1305 (m). – ^1H NMR (300 MHz): $\delta/\text{ppm} = 1.92$ –1.99 (m, 1H, CH_2CHHCHN), 2.06–2.14 (m, 1H, CH_2CHHCHN), 2.36–2.44 (m, 2H, COCH_2), 2.88 (dd, $^2J = 13.4$ Hz, $^3J = 8.5$ Hz, 1H, CHHPh), 3.23 (dd, $^2J = 13.4$ Hz, $^3J = 3.6$ Hz, 1H, CHHPh), 4.55 (tdd, $^3J = 8.5$ Hz, $^3J = 3.6$ Hz, $^3J = 1.6$ Hz, 1H, CHN), 7.28–7.44 (m, 5H, arom. H). – ^{13}C NMR (75.5 MHz): $\delta/\text{ppm} = 21.8$ (COCH_2CH_2), 31.0 (COCH_2), 39.2 ($\text{CH}_2\text{C}_{\text{ar}}$), 53.4 (OCH_3), 58.9 (CHN), 127.0 (C_{ar}), 129.4 (2 C, C_{ar}), 128.6 (2 C, C_{ar}), 136.8 ($\text{CH}_2\text{C}_{\text{ar}}$), 152.2 (COO), 173.7 (CON). – MS (EI), m/z (%): 233 (11) [M^+], 142 (100) [$\text{M}^+ - \text{C}_7\text{H}_7$], 98 (56) [$\text{C}_6\text{H}_{12}\text{N}^+$], 91 (35) [C_7H_7^+].

$\text{C}_{13}\text{H}_{15}\text{NO}_3$ Calcd: C 66.94 H 6.48 N 6.00
(233.26) Found: C 67.09 H 6.57 N 6.11.

(2R,5S)-N-Methoxycarbonyl-5-ethyl-2-methoxypyrrolidine (3aa)

Procedure C. A 1.0M solution of LiEt_3BH in THF (1.4 ml, 1.4 mmol) was added dropwise to a cooled solution of 1.32 mmol of 2-pyrrolidinone **7a** (149 mg) in 15 ml of CH_2Cl_2 at -78°C . The reaction mixture was allowed to reach 0°C , and it was subsequently stirred for 4 h at this temperature. A 1M NaOH solution (2.0 ml, 2.0 mmol) and 1.0 ml of a 35% H_2O_2 solution were added successively. The organic layer was separated and filtered through a Celite pad.

The solution so received was cooled to 0°C . 8.2 mmol of 2,2-dimethoxypropane (850 mg, 1.0 ml) and 0.1 mmol of camphor sulfonic acid (CSA) (25 mg) were added. After 30 min 10 ml of a saturated aq. NaHCO_3 solution was added. The aqueous layer was extracted with CH_2Cl_2 (3 × 10 ml). The combined organic layers were filtered through a Celite pad and concentrated *in vacuo*. The diastereomeric mixture of **3aa** was used without further purification for the transfor-

mation to 2,3-dihydropyrrolidine **7a**. Yield 139 mg (82 %). – $R_f = 0.21/0.26$ (PE/TBME = 80/20).

(2R,5S)-N-Methoxycarbonyl-5-benzyl-2-methoxypyrrolidine (3ab)

Following the protocol described in procedure C (*vide above*) 1.18 mmol 2-pyrrolidinone **7b** (275 mg) were transformed into the desired product. Yield 253 mg (86 %). – $R_f = 0.58/0.60$ (TBME).

(S)-N-Methoxycarbonyl-5-ethyl-2,3-dihydropyrrole (1aa)

Following the protocol described in procedure A (*vide infra*) 1.08 mmol of 2-methoxypyrrolidine **3aa** (139 mg) were transformed into the desired product. Purification by flash chromatography (PE/TBME = 90/10). Yield 132 mg (79 %). – $R_f = 0.39$ (PE/TBME = 80/20). – $[\alpha]_D^{25} = -145.7$ ($c = 1.6$, CHCl_3). – IR (film): $\nu/\text{cm}^{-1} = 2960$ (m), 1705 (s, C=O), 1450 (m), 1380 (m). – ^1H NMR (300 MHz): $\delta/\text{ppm} = 0.84$ (t, $^3J = 7.5$ Hz, 3H, CH_2CH_3), 1.58–1.61 (m, 2H, CH_2CH_3), 2.26 (d, $^2J = 15.8$ Hz, 1H, NCHCHH), 2.79 (dd, $^2J = 15.8$ Hz, $^3J = 1.2$ Hz, 1H, NCHCHH), 3.70 (s, 1H, OCH_3), 4.08–4.12 (m, 1H, NCHCH_2), 4.91–4.94 (m, 1H, NCHCH), 6.44–6.56 (m, 1H, NCHCH). – ^{13}C NMR (75.5 MHz): $\delta/\text{ppm} = 8.5$ (CH_2CH_3), 26.5 (CH_2CH_3), 34.2 (NCHCH_2), 52.2 (OCH_3), 58.4 (NCHCH_2), 107.2 (NCHCH), 128.8 (NCHCH). ^{13}C signal for CO was not observed. – MS (EI), m/z (%): 155 (39) [M^+], 126 (100) [$\text{M}^+ - \text{CH}_2\text{CH}_3$], 67 (45) [$\text{C}_4\text{H}_5\text{N}^+$]. – HRMS Calcd. for $\text{C}_8\text{H}_{13}\text{NO}_2$: 155.0946. Found: 155.0961.

(S)-N-Methoxycarbonyl-5-benzyl-2,3-dihydropyrrole (1ab)

Following the protocol described in procedure A (*vide above*) 1.01 mmol of 2-methoxypyrrolidine **3ab** (253 mg) were transformed into the desired product. Purification by flash chromatography (PE/TBME = 90/10). Yield 163 mg (74%). – $R_f = 0.35$ (PE/TBME = 80/20). – $[\alpha]_D^{25} = -73.5$ ($c = 0.4$, CHCl_3). – IR (film): $\nu/\text{cm}^{-1} = 2955$ (m), 1700 (vs, C=O), 1450 (m), 1385 (m). – ^1H NMR (300 MHz): $\delta/\text{ppm} = 2.33$ (d, $^2J = 15.8$ Hz, 1H, CHCHCHH), 2.66 (dd, $^2J = 13.1$ Hz, $^3J = 9.1$ Hz), 3.10 (d, $^2J = 15.8$ Hz, 1H, CHCHCHH), 3.75 (s, 3H, OCH_3), 4.31–4.43 (m, 1H, NCHCH_2), 4.86–4.95 (m, 1H, NCHCH), 6.44–6.57 (m, 1H, NCHCH), 7.18–7.31 (m, 5H, arom. H). – ^{13}C NMR (75.5 MHz): $\delta/\text{ppm} = 33.7$ (CHCHCH_2), 39.2 ($\text{CHCH}_2\text{C}_{\text{ar}}$), 52.3 (OCH_3), 58.4 ($\text{CH}_2\text{C}_{\text{ar}}$), 107.1 (NCHCH), 126.3 (C_{ar}), 128.3 (2 C, C_{ar}), 128.5 (NCHCH), 129.5 (2 C, C_{ar}), 137.7 ($\text{CH}_2\text{C}_{\text{ar}}$). ^{13}C signal for CO was not observed. – MS (EI), m/z (%): 217 (11) [M^+], 126 (100) [$\text{M}^+ - \text{C}_7\text{H}_7$], 91 (51) [C_7H_7^+].

$\text{C}_{13}\text{H}_{15}\text{NO}_2$ Calcd: C 71.87 H 6.96 N 6.45
(217.26) Found: C 72.03 H 7.06 N 6.67.

References

- [1] T. Shono, Y. Matsumura, K. Tsubata, Y. Sugihara, S.-I. Yamane, T. Kanazawa, T. Aoki, *J. Am. Chem. Soc.* **1982**, *104*, 6697
- [2] M. J. S. Carpes, P. C. M. L. Miranda, C. R. D. Coreia, *Tetrahedron Lett.* **1997**, *38*, 1869
- [3] L. F. Tietze, R. Ferraccioli, *Synlett* **1998**, 145
- [4] a) T. Bach, *Angew. Chem.* **1996**, *108*, 976; *Angew. Chem.*

- Int. Ed. Engl. **1996**, 35, 884; b) T. Bach, H. Brummerhop, *Angew. Chem.* **1998**, 110, 3577, *Angew. Chem. Int. Ed. Engl.* **1998**, 37, 3400
- [5] a) K. Nilsson, A. Hallberg, *J. Org. Chem.* **1990**, 55, 2464; b) F. Ozawa, T. Hayashi, *J. Organomet. Chem.* **1992**, 428, 267; c) F. Ozawa, Y. Kobatake, T. Hayashi, *Tetrahedron Lett.* **1993**, 34, 2505
- [6] a) Y. Nomura, K. Ogawa, Y. Takeuchi, S. Tomoda, *Chem. Lett.* **1977**, 693; b) G. A. Kraus, K. Neuenschwander, *J. Org. Chem.* **1981**, 46, 4791
- [7] R. K. Dieter, R. R. Sharma, *J. Org. Chem.* **1996**, 61, 4180
- [8] H. Dhimane, C. Vanucci-Bacqu , L. Hamon, G. Lhommet, *Eur. J. Org. Chem.* **1998**, 1955
- [9] Review: H. de Koning, W. N. Speckamp in *Methoden der Organischen Chemie (Houben-Weyl)* 4. Aufl. (Eds.: G. Helmchen, R. W. Hoffmann, J. Mulzer, E. Schaumann), Thieme, Stuttgart 1995, Vol. E 21, 1952
- [10] a) P. G. Gassman, S. J. Burns, *J. Org. Chem.* **1988**, 53, 5574; b) P. G. Gassman, S. J. Burns, K. B. Pfister, *J. Org. Chem.* **1993**, 58, 1449
- [11] a) **3a**: ref. [1]; b) **3b**: T. Nagasaka, H. Tamano, T. Maekawa, F. Hamaguchi, *Heterocycles* **1987**, 26, 617; c) **3c**: M. Pichon, B. Figad re, A. Cav , *Tetrahedron Lett.* **1996**, 37, 7963
- [12] J.  hman, P. Somfai, *Tetrahedron* **1992**, 43, 9537
- [13] H. Brummerhop, Diplomarbeit, Universit t Marburg 1997
- [14] a) **6a**: J. L. Marco, *J. Heterocycl. Chem.* **1986**, 23, 287; b) **6b**: J. Ackermann, M. Matthes, C. Tamm, *Helv. Chim. Acta* **1990**, 73, 122
- [15] H.-D. Arndt, K. Polborn, U. Koert, *Tetrahedron Lett.* **1997**, 38, 3879
- [16] W. C. Still, M. Kahn, A. J. Mitra, *J. Org. Chem.* **1978**, 43, 2923

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